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TITLE : IRON AND PHOSPHORUS ELECTROPLATING BATH

ABSTRACT : PROBLEM TO BE SOLVED: To obtain a plating film excellent in resistance to seizure and wear by adding a specified amt. of dodecyl sulfate to an iron and phosphorus electroplating bath contg. iron(II) ion and hypophosphorous acid or its salt.

SOLUTION: Dodecyl sulfate is added by  $\geq 0.5\text{g/l}$ , preferably  $0.5\text{-}1\text{g/l}$ , to an iron and phosphorus electroplating bath contg.  $20\text{-}80\text{g/l}$  iron(II) ion using ferrous sulfate, etc., as its source,  $0.05\text{-}20\text{g/l}$  water-soluble phosphorus source of hypophosphorous acid and/or hypophosphite and further contg., as required,  $0\text{-}200\text{g/l}$  ammonium sulfate as a conductive salt,  $0\text{-}60\text{g/l}$  boric acid, etc., as a pH buffer and  $0\text{-}200\text{g/l}$  ammonium bifluoride as a complexing agent. The water-soluble salts of Na, K, Li, etc., are used as the surfactant dodecyl sulfate. The plating soln. is preferably controlled to pH  $0\text{-}3.5$ , and plating is preferably conducted in this bath at room temp. to  $80^\circ\text{C}$  and at  $\geq 0.5\text{A/dm}^2$  current density.

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